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NH3 has 4 negative centres of charge/three bonding pairs and one lone pair and BF3 has 3 negative centres of charge/three bonding pairs / OWITE; (bond angles) 107° in NH3 and 120° in BF3; [4] Accept 107.5° for NH 3. (iv) BF3 not polar as no net dipole moment / BF bond polarities cancel each other out / symmetrical distribution of charge;

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